INSTRUCTIONS TO CANDIDATES

Please read this page carefully, but do not open the question paper until you are told that you may do so.

This paper is Section 2 of 3. Your supervisor will collect this question paper and answer sheet before giving out Section 3.

A separate answer sheet is provided for this section. Please check you have one. You also require a soft pencil and an eraser.

Please complete the answer sheet with your:

- BMAT candidate number
- Centre number
- Date of birth
- Name

Speed as well as accuracy is important in this section. **Work quickly, or you might not finish the paper.** There are no penalties for incorrect responses, only marks for correct answers, so you should attempt all 27 questions. Each question is worth one mark.

Answer on the sheet provided. Questions ask you to show your choice between options by shading one circle. If you make a mistake, erase thoroughly and try again.

You must complete the answer sheet within the time limit.

You can use the question paper for rough working or notes, but **no extra paper** is allowed.

Calculators are NOT permitted.

Please wait to be told you may begin before turning this page.

This question paper consists of 17 printed pages and 3 blank pages.
1 Enzymes are biological catalysts involved in many reactions in cells.

Which of the following could affect an enzyme's activity?

1 Treating the enzyme with protease before use.
2 Boiling the enzyme before use.
3 Using an enzyme warmed to room temperature after it has been stored in a refrigerator.

A none of them
B 1 and 2 only
C 1 and 3 only
D 2 and 3 only
E 1, 2 and 3

2 Which of the following could possibly take part in an addition polymerisation reaction?

1 CHI₃
2 C₂₄H₄₈
3 C₃H₇Br
4 C₄H₆Cl₂
5 C₈H₁₂Cl₄

A 1, 2 and 3 only
B 1, 2 and 4 only
C 1, 3 and 5 only
D 2, 3 and 4 only
E 2, 4 and 5 only
F 3, 4 and 5 only
After running a marathon, a particular runner wraps herself in a thin aluminium-covered plastic sheet in order to keep warm.

Here are some suggested statements about the effects this may have on heat loss from her body:

1. There is less heat loss through conduction, because aluminium is a good conductor of heat.
2. There is less heat loss through convection, because air is trapped between the sheet and the body.
3. There is less heat loss through radiation, because shiny metal surfaces are poor emitters of thermal radiation.

Which of these statements is/are correct?

A none of them
B 1 only
C 2 only
D 3 only
E 1 and 2 only
F 1 and 3 only
G 2 and 3 only
H 1, 2 and 3
The right-angled triangle shown has horizontal and vertical sides measuring \((4 + \sqrt{2})\) cm and \((2 - \sqrt{2})\) cm respectively.

[diagram not to scale]

Calculate the area of the triangle.

A \((3 - 3\sqrt{2})\) cm\(^2\)

B \((3 + 3\sqrt{2})\) cm\(^2\)

C \((3 - \sqrt{2})\) cm\(^2\)

D \((5 - \sqrt{2})\) cm\(^2\)

E \((5 - 3\sqrt{2})\) cm\(^2\)

F \((5 + 3\sqrt{2})\) cm\(^2\)

One healthy male human testis cell undergoes meiosis.

How many chromosomes, in total, are present after meiosis is completed?

A 23

B 44

C 46

D 92

E 184
6  Inorganic substances are put into detergents to act as a buffer, i.e. to keep the pH close to 7. Sodium tripolyphosphate is one major one that is used. The equation for its formation is shown below:

\[ a\text{NaH}_2\text{PO}_4 + b\text{Na}_2\text{HPO}_4 \rightarrow c\text{Na}_5\text{P}_3\text{O}_{10} + d\text{H}_2\text{O} \]

What are the values of \(a\), \(b\), \(c\) and \(d\), respectively?

A  3, 1, 1, 3  
B  6, 2, 2, 7  
C  2, 4, 2, 2  
D  1, 2, 1, 2  
E  2, 2, 1, 2

7  Two identical cars, P and Q, start at the same level. Car P moves at a constant speed of \(10 \text{ m s}^{-1}\) up a hill to a height of 25 m in a time of 20 s. In the same time, car Q moves at a constant speed of \(20 \text{ m s}^{-1}\) up a hill to a height of 50 m.

Which of the following statements are correct for the kinetic energies of the cars while they are travelling up the hills, and for their gravitational potential energies once they are at the top?

<table>
<thead>
<tr>
<th></th>
<th>kinetic energy</th>
<th>gravitational potential energy</th>
</tr>
</thead>
<tbody>
<tr>
<td>A</td>
<td>car Q has twice as much as car P</td>
<td>car Q has twice as much as car P</td>
</tr>
<tr>
<td>B</td>
<td>car Q has twice as much as car P</td>
<td>car Q has four times as much as car P</td>
</tr>
<tr>
<td>C</td>
<td>car Q has four times as much as car P</td>
<td>car Q has twice as much as car P</td>
</tr>
<tr>
<td>D</td>
<td>car Q has four times as much as car P</td>
<td>car Q has four times as much as car P</td>
</tr>
</tbody>
</table>

[diagram not to scale]
Simplify:

\[
\left( \frac{\frac{3}{2x^2 y^3}}{\sqrt{z}} \right)^2
\]

A \quad \frac{4x^6 y^5}{z^2}

B \quad \frac{4x^3 y^6}{z}

C \quad \frac{4x^3 y^5}{z}

D \quad \frac{5x^2 y^5}{z}

E \quad \frac{5x^2 y^5}{z^2}
Individual P in the family pedigree below is homozygous dominant and individual Q is homozygous recessive for a particular feature.

What is the percentage probability that individual U is homozygous recessive if:

<table>
<thead>
<tr>
<th></th>
<th>i) T is homozygous recessive</th>
<th>ii) T is heterozygous</th>
</tr>
</thead>
<tbody>
<tr>
<td>A</td>
<td>50%</td>
<td>0%</td>
</tr>
<tr>
<td>B</td>
<td>25%</td>
<td>50%</td>
</tr>
<tr>
<td>C</td>
<td>50%</td>
<td>25%</td>
</tr>
<tr>
<td>D</td>
<td>25%</td>
<td>0%</td>
</tr>
<tr>
<td>E</td>
<td>0%</td>
<td>25%</td>
</tr>
</tbody>
</table>
10 The reaction between nitrogen and hydrogen to form ammonia is exothermic.

\[ \text{N}_2(\text{g}) + 3\text{H}_2(\text{g}) \rightleftharpoons 2\text{NH}_3(\text{g}) \quad \Delta H \text{ is negative} \]

The bond energies in the three molecules are as shown.

- N≡N \( x \text{ kJ mol}^{-1} \)
- H–H \( y \text{ kJ mol}^{-1} \)
- N–H \( z \text{ kJ mol}^{-1} \)

Which statement can be deduced from this information?

A \( z > x + y \)
B \( 2z > x + y \)
C \( 2z > x + 3y \)
D \( 6z > x + 3y \)

11 Following a natural disaster, a container of emergency supplies is dropped from a hovering (stationary) helicopter to people on the ground. The container drops vertically until it reaches its terminal speed. A parachute attached to the container then opens and the container quickly reaches a new terminal speed.

Which graph represents how the speed of the container changes with time from when it is released from the helicopter to just before it hits the ground?

A

B

C

D

E

F
12 A bag contains $x$ red balls, $y$ blue balls and $z$ yellow balls. One ball at random is taken out and put back. A second ball at random is taken out and put back.

If the balls are identical in all respects except colour and are well mixed, what is the probability that the first ball was red and the second blue?

A \( \frac{(x + y)}{(x + y + z)} \)

B \( \frac{(x + y)}{(x + y + z)^2} \)

C \( \frac{xy}{(y + z)(x + z)} \)

D \( \frac{xy}{(x + y + z)(x + z)} \)

E \( \frac{xy}{(x + y + z)^2} \)

13 Signals travelling along a reflex arc pass from one neuron to the next neuron by the release of transmitter molecules. The statements below are about this process.

1 The signal is transmitted across the synapse by osmosis.

2 Transmitter molecules are released once the signal has been transmitted across the synapse.

3 The release of transmitter molecules is triggered by the signal.

4 The signal is transmitted across the synapse by diffusion.

Which of the above statements are correct?

A 1 and 2 only

B 1 and 3 only

C 1 and 4 only

D 2 and 4 only

E 3 and 4 only
Hydrogen, magnesium and phosphate can exist as the following ions: H⁺, Mg²⁺, PO₄³⁻.

Different salts can be formed from these ions.

Which formula below is a formula for one of these salts?

A  Mg(HPO₄)₂
B  Mg(H₂PO₄)₂
C  MgH₂PO₄
D  Mg(H₃PO₄)₂
E  Mg₂HPO₄
F  Mg₂H₂PO₄

A radioactive isotope X, which is used in hospitals, has a half-life of 6.0 hours. It decays to form an isotope Y of a different element which is not radioactive.

A pure sample of X is produced in a nuclear reactor. Twelve hours later, when the sample arrives at a hospital, the number of atoms of X in the sample is $1.6 \times 10^{23}$.

How many atoms of X were in the original pure sample, and how many atoms of Y will the sample contain exactly 24 hours after it arrives at the hospital?

<table>
<thead>
<tr>
<th>original number of atoms of X</th>
<th>number of atoms of Y after 24 hours</th>
</tr>
</thead>
<tbody>
<tr>
<td>A  $3.2 \times 10^{23}$</td>
<td>1.0 $\times 10^{22}$</td>
</tr>
<tr>
<td>B  $3.2 \times 10^{23}$</td>
<td>2.0 $\times 10^{22}$</td>
</tr>
<tr>
<td>C  $3.2 \times 10^{23}$</td>
<td>3.0 $\times 10^{23}$</td>
</tr>
<tr>
<td>D  $3.2 \times 10^{23}$</td>
<td>3.1 $\times 10^{23}$</td>
</tr>
<tr>
<td>E  $6.4 \times 10^{23}$</td>
<td>1.0 $\times 10^{22}$</td>
</tr>
<tr>
<td>F  $6.4 \times 10^{23}$</td>
<td>2.0 $\times 10^{22}$</td>
</tr>
<tr>
<td>G  $6.4 \times 10^{23}$</td>
<td>6.2 $\times 10^{23}$</td>
</tr>
<tr>
<td>H  $6.4 \times 10^{23}$</td>
<td>6.3 $\times 10^{23}$</td>
</tr>
</tbody>
</table>
In statistics, Spearman’s rank correlation coefficient is given by the formula:

\[ r = 1 - \frac{6 \sum d^2}{n(n^2 - 1)} \]

Rearrange the formula to make \( \sum d^2 \) the subject.

A \[ \sum d^2 = 1 - \frac{r(n^3 - n)}{6} \]

B \[ \sum d^2 = \left( \frac{(1 - r)(n^3 - n)}{6} \right)^2 \]

C \[ \sum d^2 = \frac{(1 - r)(n^3 - 1)}{6} \]

D \[ \sum d^2 = \frac{(1 + r)(n^3 - n)}{6} \]

E \[ \sum d^2 = \frac{(1 - r)(n^3 - n)}{6} \]

The three statements below are about breathing out.

1. The ribs swing down and inwards when breathing out.
2. The diaphragm muscles contract when breathing out.
3. The pressure in the lungs increases when breathing out.

Which of these statements is/are correct?

A 1 only
B 2 only
C 3 only
D 1 and 2 only
E 1 and 3 only
F 2 and 3 only
An ore of lead contains lead only in the form of lead(II) sulfide, PbS.

By mass, 75% of this ore is PbS.

Calculate the maximum mass of lead that can be extracted from 480 kg of the ore.

\[ A_r: \text{Pb} = 207; \ S = 32 \]

A. 48.20 kg

B. 103.9 kg

C. 180.0 kg

D. 311.8 kg

E. 415.7 kg
The depth of water in a particular tidal harbour varies with time as shown in the graph:

If the variation in depth caused by the effect of the tide is considered as a wave, what are the amplitude and frequency of this wave?

<table>
<thead>
<tr>
<th>amplitude / m</th>
<th>frequency / Hz</th>
</tr>
</thead>
<tbody>
<tr>
<td>A 3</td>
<td>( \frac{1}{12 \times 3600} )</td>
</tr>
<tr>
<td>B 3</td>
<td>( \frac{3600}{12} )</td>
</tr>
<tr>
<td>C 6</td>
<td>( \frac{1}{24 \times 3600} )</td>
</tr>
<tr>
<td>D 6</td>
<td>( \frac{3600}{24} )</td>
</tr>
<tr>
<td>E 8</td>
<td>( \frac{1}{12 \times 3600} )</td>
</tr>
<tr>
<td>F 8</td>
<td>( \frac{3600}{12} )</td>
</tr>
<tr>
<td>G 16</td>
<td>( \frac{1}{24 \times 3600} )</td>
</tr>
<tr>
<td>H 16</td>
<td>( \frac{3600}{24} )</td>
</tr>
</tbody>
</table>
The graphs of the following equations are drawn:

1. \( y = 3x - 2 \)
2. \( y = x^2 \)
3. \( y = 1 - x^2 \)
4. \( y = x + 6 \)

Which pair of graphs do not intersect?

A. 1 and 2
B. 1 and 3
C. 2 and 3
D. 2 and 4
E. 3 and 4

The table shows the concentrations, in arbitrary units, of four substances. These substances are present in each of two animal cells, 1 and 2, whose cell membranes are in contact.

<table>
<thead>
<tr>
<th>substance</th>
<th>cell 1</th>
<th>cell 2</th>
</tr>
</thead>
<tbody>
<tr>
<td>P</td>
<td>9</td>
<td>15</td>
</tr>
<tr>
<td>Q</td>
<td>12</td>
<td>8</td>
</tr>
<tr>
<td>R</td>
<td>7</td>
<td>4</td>
</tr>
<tr>
<td>S</td>
<td>6</td>
<td>3</td>
</tr>
</tbody>
</table>

Which overall movement of a substance between the two cells requires oxygen?

A. P: cell 2 → cell 1
B. Q: cell 1 → cell 2
C. R: cell 2 → cell 1
D. S: cell 1 → cell 2
22  Assume that element Y forms a naturally-occurring ion \( Y^{3-} \) with the electronic configuration 2,8,8.

Using this information, determine to which Group and Period of the IUPAC Periodic Table Y would belong.

<table>
<thead>
<tr>
<th>Group</th>
<th>Period</th>
</tr>
</thead>
<tbody>
<tr>
<td>A</td>
<td>13</td>
</tr>
<tr>
<td>B</td>
<td>13</td>
</tr>
<tr>
<td>C</td>
<td>15</td>
</tr>
<tr>
<td>D</td>
<td>15</td>
</tr>
<tr>
<td>E</td>
<td>18</td>
</tr>
<tr>
<td>F</td>
<td>18</td>
</tr>
</tbody>
</table>

23  A 100% efficient transformer has 1500 turns in its primary coil. The input to the transformer is 250 V alternating current (ac). The output is connected to a resistor. The output current is 10 A and the output power is 0.50 kW.

What is the number of turns in the secondary coil?

A  75
B  300
C  750
D  7500
E  30000
24 A solid sphere of radius \( r \) fits inside a hollow cylinder. The cylinder has the same internal diameter and length as the diameter of the sphere.

The volume of a sphere is \( \frac{4}{3} \pi r^3 \), where \( r \) is the radius of the sphere.

What fraction of the space inside the cylinder is taken up by the sphere?

A \( \frac{1}{4} \)

B \( \frac{1}{3} \)

C \( \frac{1}{2} \)

D \( \frac{2}{3} \)

E \( \frac{3}{4} \)

25 Which one of the following is not needed in order to genetically engineer bacterial cells to produce a fluorescent protein from a jellyfish?

A a plasmid or viral vector

B enzymes to cut DNA molecules

C fluorescent protein from a jellyfish

D ligase enzyme
26 A compound of iodine and oxygen contains 63.5 g of iodine and 20.0 g of oxygen.
Which one of the following is its empirical formula?
(A; I = 127; O = 16)

A  IO
B  IO₂
C  I₂O
D  I₂O₃
E  I₂O₅
F  I₅O₂

27 The microwaves generated in a microwave oven travel through air at a speed of $3.0 \times 10^8$ m s⁻¹, with a wavelength of 12 cm. They pass through plastic food containers, but at a reduced speed of $2.0 \times 10^8$ m s⁻¹.

What are the wavelength and frequency of these microwaves as they pass through a plastic food container?

<table>
<thead>
<tr>
<th>wavelength / cm</th>
<th>frequency / Hz</th>
</tr>
</thead>
<tbody>
<tr>
<td>A 8.0</td>
<td>1.7 × 10⁹</td>
</tr>
<tr>
<td>B 8.0</td>
<td>2.5 × 10⁹</td>
</tr>
<tr>
<td>C 8.0</td>
<td>3.8 × 10⁹</td>
</tr>
<tr>
<td>D 12</td>
<td>1.7 × 10⁹</td>
</tr>
<tr>
<td>E 12</td>
<td>3.8 × 10⁹</td>
</tr>
<tr>
<td>F 18</td>
<td>1.7 × 10⁹</td>
</tr>
<tr>
<td>G 18</td>
<td>2.5 × 10⁹</td>
</tr>
<tr>
<td>H 18</td>
<td>3.8 × 10⁹</td>
</tr>
</tbody>
</table>

END OF TEST
This document was initially designed for print and as such does not reach accessibility standard WCAG 2.1 in a number of ways including missing text alternatives and missing document structure.

If you need this document in a different format please email admissionstesting@cambridgeassessment.org.uk telling us your name, email address and requirements and we will respond within 15 working days.